

Electrochemical Gas Cells in Air Quality Networks- Solving the Problems



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Who is Alphasense?

Private UK limited company, 10 minutes from Stansted airport
Started January 1997

65 people: 15 Technical, 40 manufacturing

Significant R&D investment, mostly with UK Universities

Markets: Industrial Safety and Air Quality gas sensors

Sensor technologies: electrochemical, metal oxides,

Optical particle counting, NDIR, PID, MEMS, spectroscopy

We make gas sensors and PM_{2.5}/10 OPCs. That's all.

Air Quality standards differ in USA and Europe, but **we want to resolve 10-20 ppb with an error less than 20-50 ppb**

| Pollutant | USA | | Europe | | EN standard |
|----------------------------------|-------------------------|-----------------|-------------------------|------------------|-------------|
| | ppb | Period | ppb | Limit | |
| Ozone | 75 | 8 hour average | 120 | Alert threshold | EN 14625 |
| NO _x /NO ₂ | 50 | Annual mean | 210 | Alert threshold | EN 1421 1 |
| NO _x /NO ₂ | 100 | 1 hour mean | 105 | 1 hr limit value | EN 1421 1 |
| SO ₂ | 500 | 3 hour mean | 200 | Alert threshold | EN 1421 2 |
| SO ₂ | 140 | 24 hour mean | 140 | 1 hr limit value | EN 1421 2 |
| CO | 35 ppm | 1 hour average | 8 ppm | 8 hr limit value | EN 14626 |
| | µg/m³ | | µg/m³ | | |
| PM 10 | 150 | 24 hour average | 40 | Annual mean | EN 12341 |
| PM 2.5 | 15 | Annual | 25 | Annual mean | EN 14907 |

Most difficult targets to measure accurately: NO₂ and PM 2.5

Electrochemical amperometric gas sensors

Used in the AQMesh pods, MESSAGE and Heathrow projects

Low cost

Very low power

now- LM91000 smart chip

PID for VOCs

SO₂

CO

NO₂



NO

O₃



28 x 74 mm PCB, 2mA/ 3.6V power

B series sensors for fixed site networks

32 mm dia



A series sensors for mobile networks

Inside a 4- electrode amperometric electrochemical gas sensor

Working electrode

Counter electrode

Reference electrode

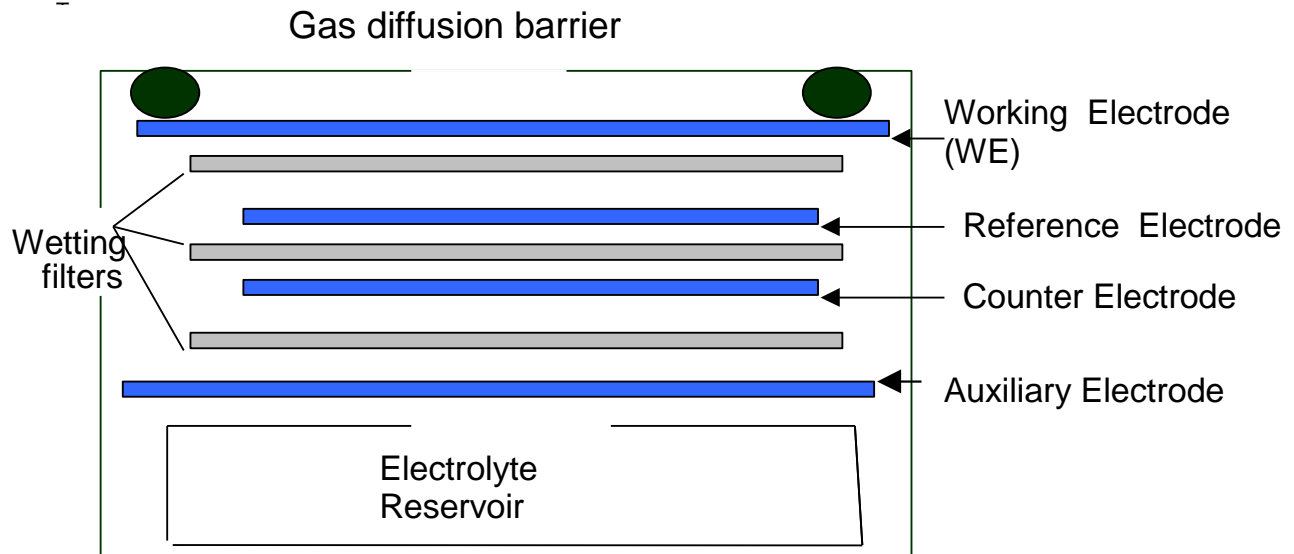
Auxiliary electrode

analyte reaction: oxidation or reduction

balances the Working electrode reaction

sets the WE potential for **selectivity**

corrects the zero current of the WE



Required sensor performance for Air Quality networks

Three requirements-

ppb **Sensitivity**/ Limit of Detection

Long term zero and sensitivity **Stability**

Selectivity, removing interfering gases

Gas Sensor Limit of Detection

(2 sd as the concentration approaches 0 ppb)

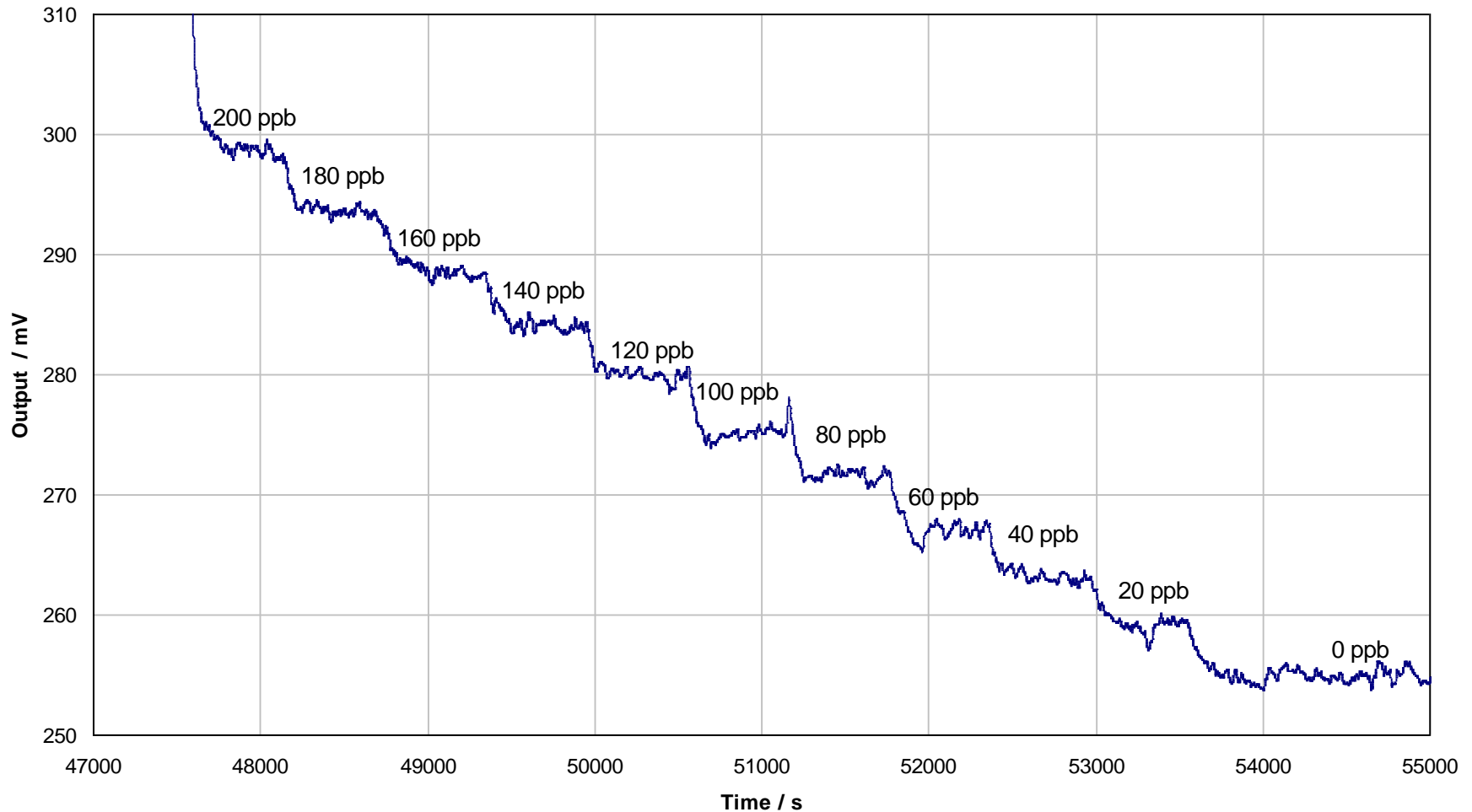
| LoD (ppb) | CO | H₂S | NO | NO₂ | O₃ | SO₂ |
|------------------|-----------|-----------------------|-----------|-----------------------|----------------------|-----------------------|
| A series | 20 | 5 | 80 | 15 | 5 | 15 |
| B series | 4 | 1 | 15 | 12 | 4 | 5 |
| | | | | | | |

Mobile A series sensors have higher limit of detection, but this is being improved with quieter electronics (esp. NO)

NO2-A4

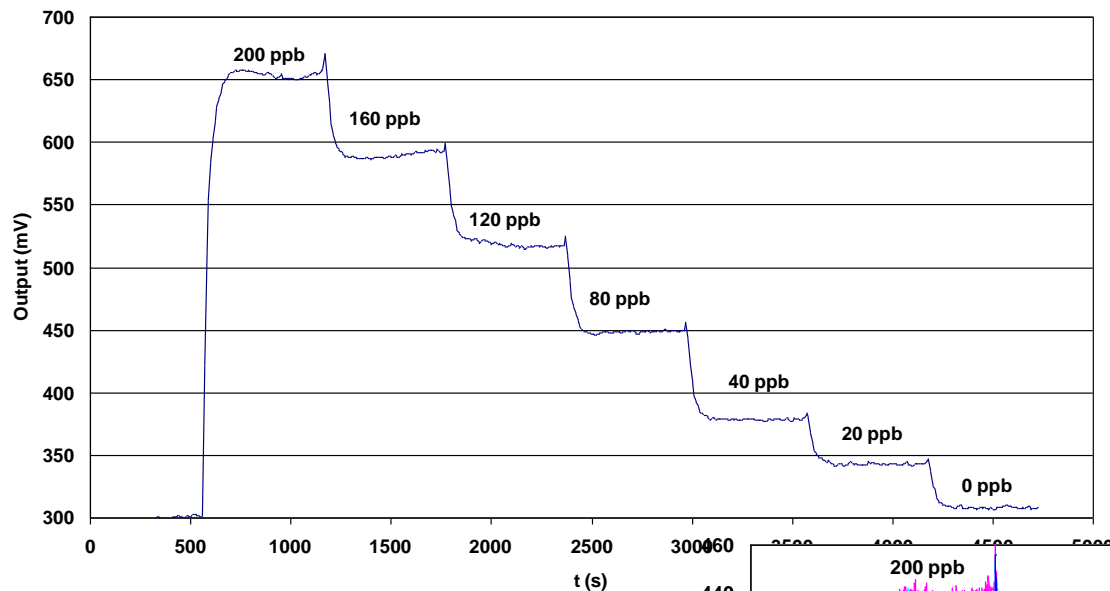
0-200 ppb linearity test (raw data)

8/05/13 CO/NO₂/O₃ triple A4 ISB : NO₂ calibration



Some examples of low level sensitivity

H2S-B4

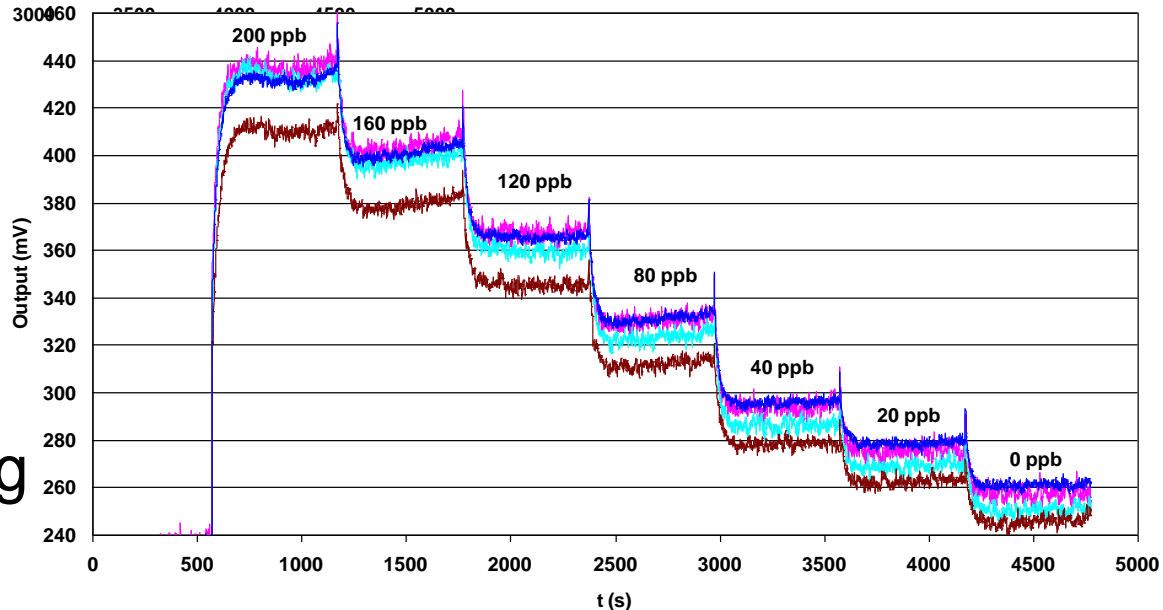


H2S-B4

(for odour detection)
shows 1 ppb noise!

Data before smoothing

H2S-A4

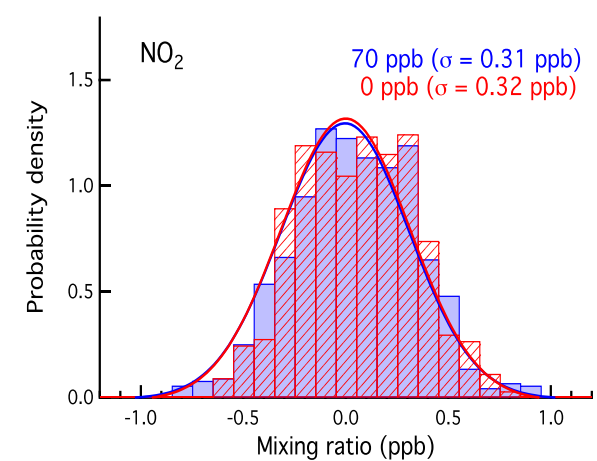
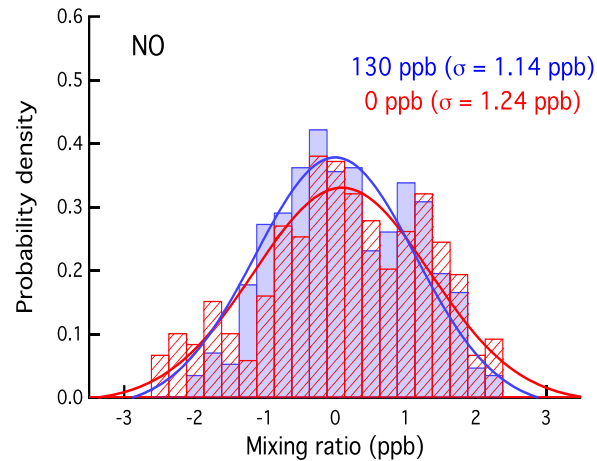
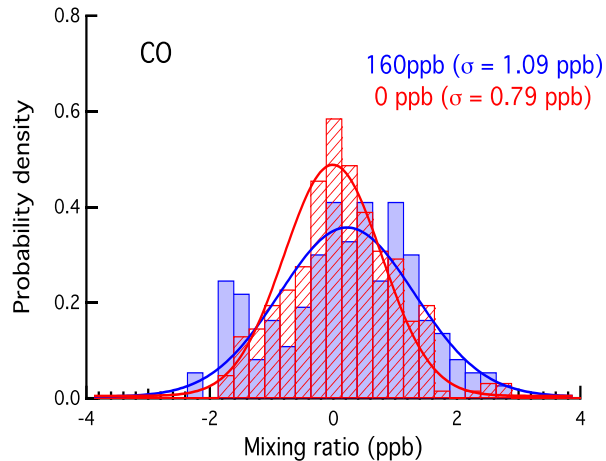


H2S-A4 has higher
noise- but still
5 ppb resolution

Data before smoothing

Cambridge laboratory measurements define noise at 1-2 ppb after smoothing.

Noise characteristics:



- Typical sensor sensitivities/LoD are < 5 ppb ($< 7 \mu\text{g}/\text{m}^3$) for CO, 1-2 ppb ($\sim 2-4 \mu\text{g}/\text{m}^3$) for NO and NO₂.
- SO₂, O₃ have comparable performance to NO_x.
- Typical sensor $t_{90} \sim 10-20$ s (determined by diffusion)

Zero baseline stability and correction

Is the zero baseline stable over time? If not, then the calculated absolute concentration using baseline correction will drift.

The zero baseline also changes with temperature. At low temperatures we are fine, but above 25°C then the correction algorithm must be good.

How to correct?

- Using **knowledge of atmospheric chemistry**
- Using **statistics and oversampling**: used by some users
- **Electrochemistry**: the underpinning process of the sensors- our method

Electrochemical correction of Zero current:

use the 4th electrode

zero current temperature dependence is very different (x10, x100) for each sensor.

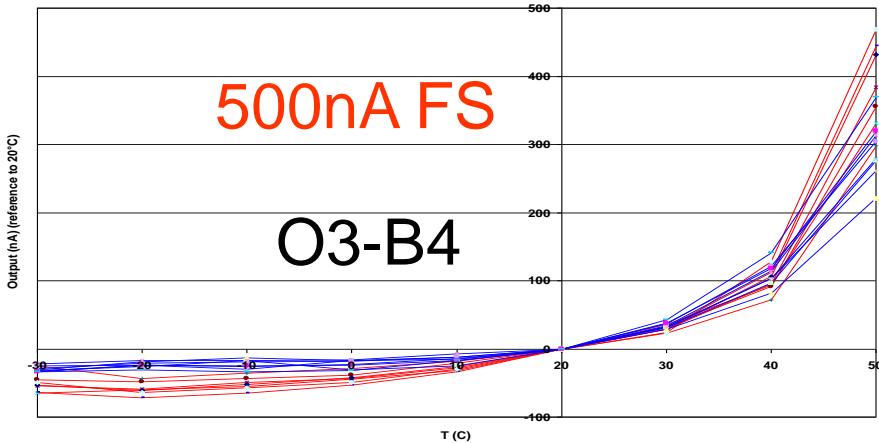
Follows Arrhenius behaviour, so we can model and predict.

Red: working electrode
Blue: auxiliary electrode

O3-B4

500nA FS

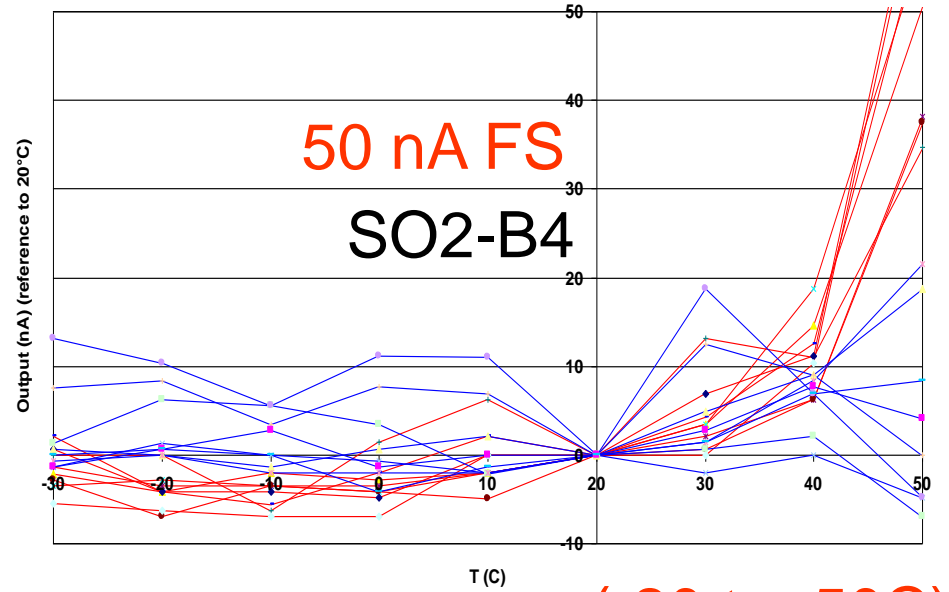
O3-B4



SO2-B4

50 nA FS

SO2-B4

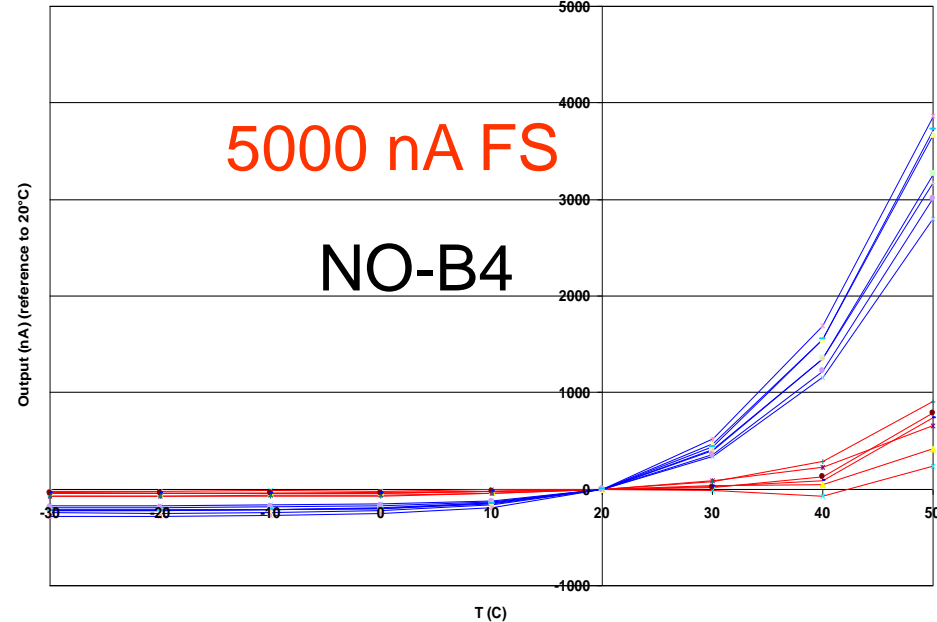


(-20 to +50C)

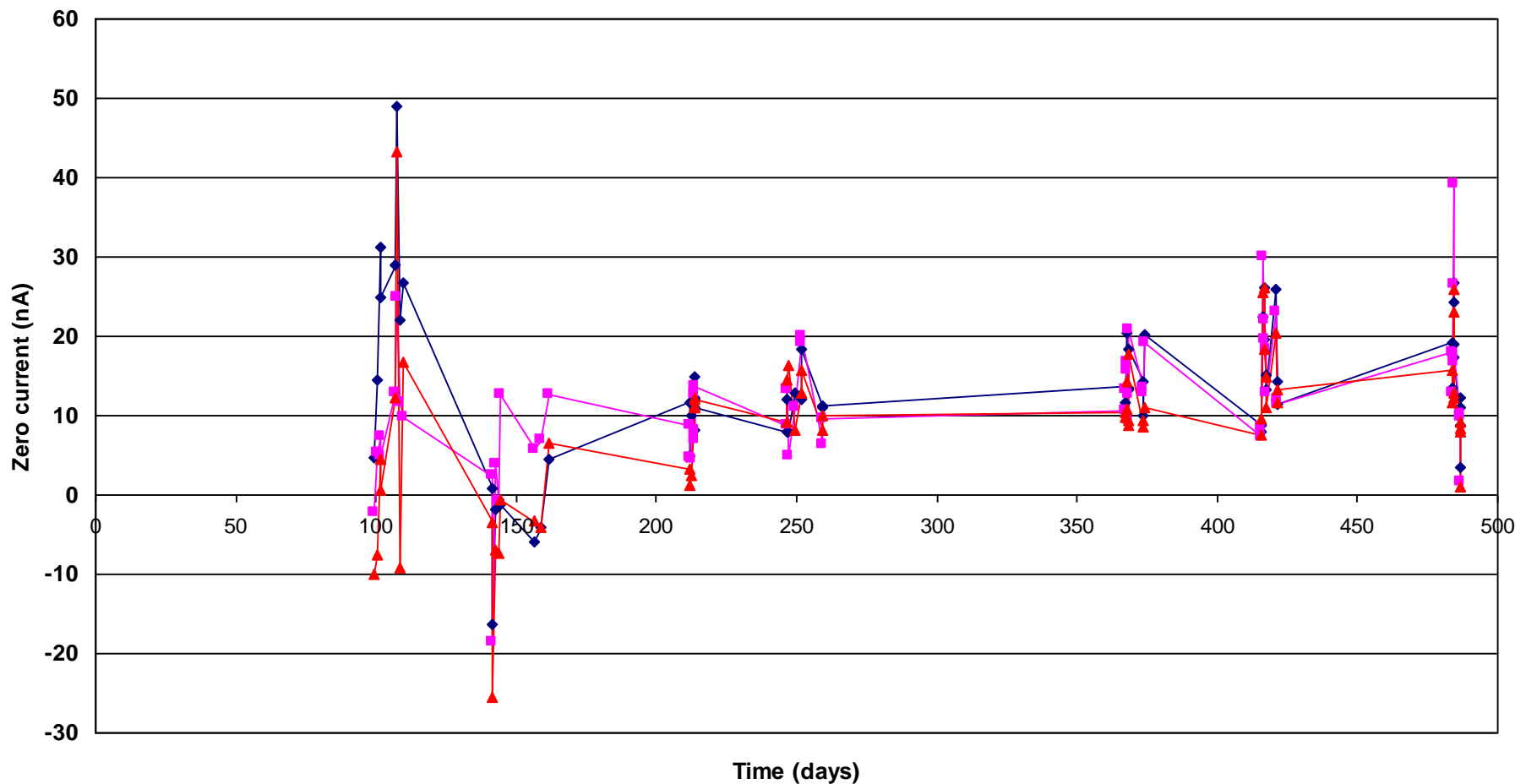
NO-B4
Working and Auxiliary Electrode

5000 nA FS

NO-B4



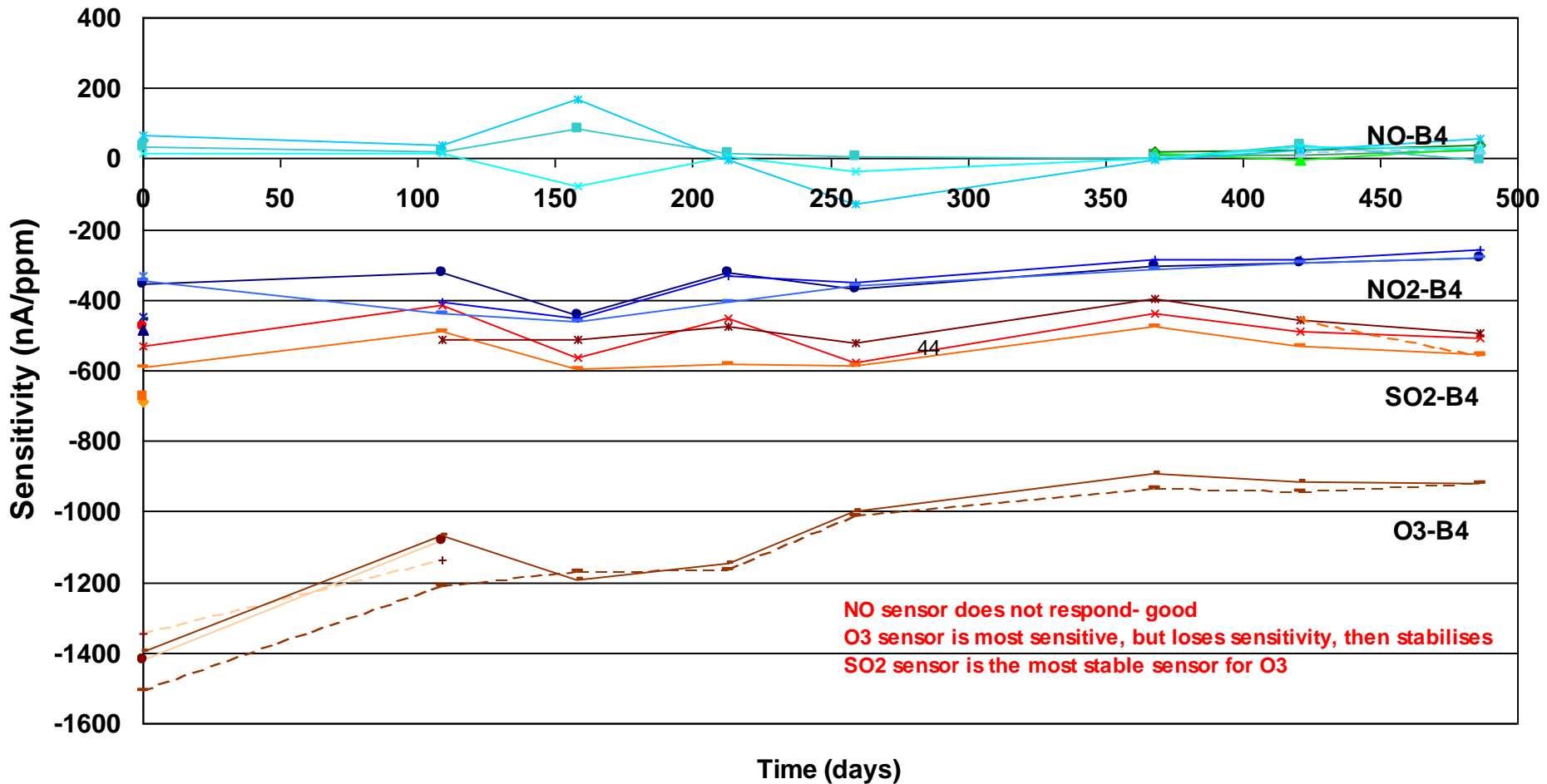
NO2-B4 zero current shows reasonable stability over 500 days ± 12 nA (equivalent to ± 20 ppb)



Stability to 100 ppb ozone

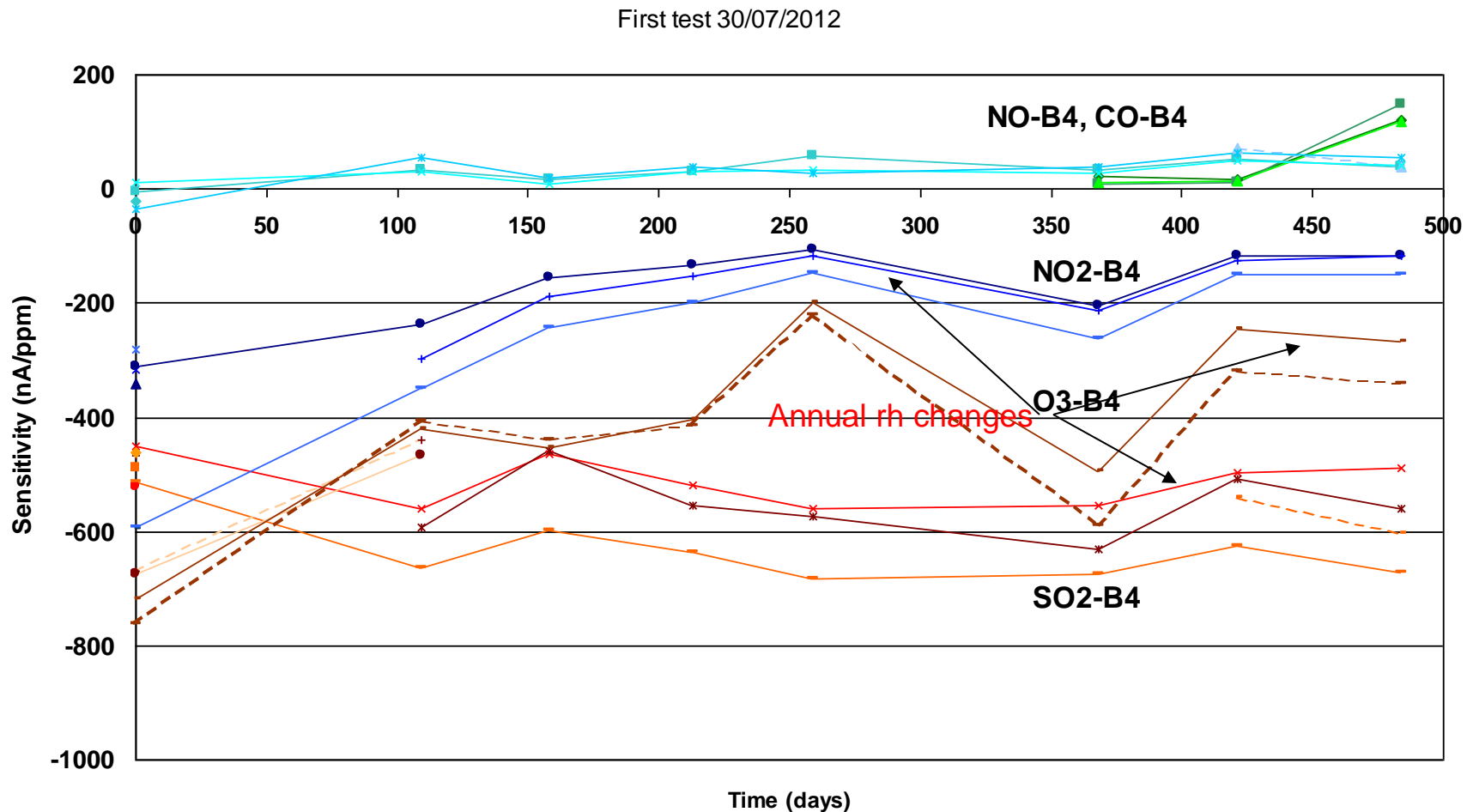
Response to 100ppb O₃

First test 30/07/2012



Stability to 350 ppb NO₂

Better stability after first 100 days, BUT: You need to stabilise in the environment the sensors will be operating



NO₂ and ozone

Difficult to separate: both are powerful oxidants

Separation Methods:

- 1 use multiple sensors: you cannot measure two parameters with one sensor. We recommend 3 sensors.
- 2 Ensure the x-sensitivites are stable and use the small ratio differences to deconvolve.
- 3 Use chemical filters to remove, for example the ozone from NO₂ sensor- so one summed sensor, one speciated sensor.

Conclusions

- Low power, low cost electrochemical sensors can measure low ppb concentrations of inorganic gases.
- Wireless air quality networks are not replacing AURNs, but are adding the extra data for research, filling in fine detail, locating emission sources and enticing the citizen to get involved.
- Our work is not done: better speciation, improving stability and modelling sensors (both transport and chemistry/ electrochemistry) are our continuing work.
- VOCs and PM are necessary partners in any complete air quality network.

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**Thank you
and have a
Merry
Christmas**

**HM Queen Elizabeth II,
patroness of Queens' College**

Any questions?

