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An account of our efforts towards air quality monitoring in epitaxial graphene on SiC



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Outline

- > Why graphene sensors?
- > Epitaxial graphene on SiC
- > Effect of graphene layer thickness on gas sensitivity and selectivity
- Understanding unintentional doping in epitaxial graphene
- Controlling graphene layer uniformity
- Tuning sensor properties by surface modifications



Why Graphene sensors?

> Unique band structure of graphene leads to a low density of states near the Dirac point (E_D) – small changes in the number of charge carriers results in large changes in the electronic state

Every atom at the surface – ultimate surface to volume ratio

Low noise, chemically stable (in non-oxidizing environment)

Graphene is highly sensitive to chemical gating due to its linear energy dispersion and vanishing density of states near the Dirac point and therefore has potential as a low noise, ultra-sensitive transducer.

Graphene sensors are normally highly sensitive, but suffer from poor reproducibility, selectivity, and speed of response....

Reproducibility is an issue that partly arises from the graphene synthesis











Graphen

manufactures and supplies

Graphene on SiC

Spin off from Linköping University, Sweden

22.11.2011

Sublimation of Si from SiC in Ar at 2000°C

Scalable, wafer-scale films compatible with standard semiconductor processing

High thickness uniformity (> 90% ML, rest 2 ML)

> Thickness controlled by temperature

Silicon carbide

Ceramic

□High chemical inertness

- Oxidation resistant
- Stable at high temperature
- Hardness
- □ Melting point ~ 2700°C
- Light-weight



Semiconductor

- □ Wide band gap
- □ High electron drift velocity
- High breakdown field
- High thermal conductivity

Polytypism: > 200 chemically identical polytypes



Graphene production



S. Sonde et al., Physical Review B 80, 241406 (R) (2009)



Sensor response to environmental gating

NO₂ strongly electron withdrawing



Why is single layer more sensitive?

- Current flow through all layers gas adsorption only on top layer
- Different band structure leads to different responsivity; change resistivity with carrier density
- Or difference in sticking coefficients of gases on single and multi layer graphene

R. Pearce et al. Sens. and actuators B. Chem., 155(2): 451-455, 2011



NO₂ sensing, single or double layer graphene?

Scanning Kelvin probe microscopy – work function mapping

Nanoscale mapping of graphene thickness uniformity and doping



□ Shifts from single to bilayer domains normally occur at terrace edges

 $\rightarrow \Delta \Phi$ between 1LG and 2LG allows nanoscale mapping of graphene thickness

600 900 1200 1500

≈ 50 mV

Distance (nm)

300

0

> Controllable environment allows observing changes in 1LG and 2LG upon gas interaction Eriksson et al., Applied Physics Letters 100 (2012) 24160



∆V (mV) 20-10**Potential distribution**

1L

≈ 88 % 1L

10 20 30 40 50 60

 $\Delta V (mV)$

2L

20 18

16 14

12

-20 -10 0

Counts

SKPM in controlled environment





In N₂: after vacuum $\Phi_{1LG} = \Phi_{2LG}$



Corrugations in 2LG upon repeated gas exposure and vacuum 'cleaning'

s exposure and



NO₂: Electron withdrawing gas increases ΔV_{CPD} , _{2L-1L}



 V_{CPD} (1LG) and V_{CPD} (2LG) decrease, but V_{CPD} (1LG) decreases more



Different shifts for 1LG and 2LG?



- Linear for 1LG
- Parabolic for 2LG



From 1-2L ΔV_{CPD}: Non-invasive estimation of carrier concentration

(1)
$$\Delta n \ 1\text{LG} = \frac{2e \ \partial V_{CPD}\sqrt{n}}{\hbar v_F \sqrt{\pi}} - \frac{(e \ \partial V_{CPD})^2}{\hbar^2 v_F^2 \pi}$$

(2)
$$\Delta n \ 2\text{LG} = \frac{\delta V_{CPD} \ e \ 2m^*}{\hbar^2 \pi}$$

Response to < 1 ppm NO₂ vs. time

R. Pearce, J. Eriksson, T. Iakimov, L. Hultman, A. Lloyd Spetz and *R. Yakimova, ACS Nano 7 (5), pp 4647–4656 (2013)*

- Calculated change in carrier concentration not the same for 1 and 2LG
- Different responsivity for 1 and 2LG doesn't account for all difference in sensitivity
- Different sticking coefficients also important

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Effect of humidity on surface potential

- Environment affects the surface potential
- V_{CPD} (1LG) decreases
- V_{CPD} (2LG) constant



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Controlling graphene layer uniformity and unintentional doping

□ Large spread observed in N_s for samples grown under identical conditions

□ There is strong indication that a correlation exists between the substrate surface morphology and the electronic properties of the epitaxial graphene.

Yakes et al., Nano Lett. 10, 1559–1562 (2010)



Mono layer coverage depends on terrace width



□ Terrace width < 300 nm – no 1LG

□ As the terrace width increases, the area covered by 1LG increases

 \Box Graphene growth starts at step edges; many step edges \rightarrow many nucleation sites

□ Terrace width > 1200 nm – gradual decrease of 1LG - Island growth in the absence of steps

□ Substrate polytype and doping for hexagonal SiC (n-type 6H-SiC or SI 4H-SiC) do not significantly influence uniformity

□3C-SiC – higher 1LG % for lower terrace width , 1LG % independent on terrace width

Carrier concentration depends on SiC surface



Surface restructuring during Si sublimation



- All substrates undergo significant restructuring during graphene growth
 - **Differing restructuring of different nominally on-axis SiC substrates**
 - No correlation seen between SiC step distance before growth and how much the SiC restructures upon graphene growth

3C-SiC restructures less, and even a reduction of the terrace width is possible

Effects of surface restructuring

30



More significant restructuring leads to less uniform graphene

Work function decreases (n-doping increases) with amount of restructuring

Minimize the restructuring → Use 3C-SiC substrates

Due to less step-bunching, 3C-SiC better lends itself to a well-controlled surface morphology and better control of the electronic properties of the graphene

Eriksson et al., Mater. Sci. Forum 740-742 (2013) 153-156



Step bunching in 4H, 6H, and 3C-SiC





Yazdi et al., Carbon 57 477 –484 (2013)



Uniform 1LG leads to very reproducible sensor characteristics



ΔS depends on thickness due to differing

Uniform 1LG leads to very reproducible sensor characteristics



1LG is more sensitive to NOx than 2LG or MLG

Uniform 1LG required for maximum sensitivity and reproducibility

1LG show identical response

reproducible sensor fabrication

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Graphene sensors issues: selectivity, response/recovery time, reproducibility

- Obstacles: sensitivity, selectivity, response/recovery time, reproducibility
 - Sensitivity
 - UV "cleaning": ppt and ppq level detection surpassing specially trained dogs!
 - O-functionalizes graphene

Chen et al., Applied Physics Letters **101**, 053119 (2012) *Sensors and Actuators B* **166–167**, 172–176 (2012)

Selectivity

Surface functionalizations by e.g. oxygen, nanoparticles, defect engineering, smart operation and smart analysis
J. Mater. Chem. 22, 11009 (2012)

Response/recovery times

Functionalization, current-, bias- or temperature cycling. Integration of UV-LED

Reproducibility

- Sensors on epitaxial graphene
- Variations in Ns can be compensated by the use of FET sensor
 - SenSiC AB patent application







Functionalization with metal and metal oxide nanostructures for selectivity tuning



Aim: To develop a reproducible method for functionalization with nano structures

- Thin layers of Au and Pt DC sputtered onto EG/SiC at elevated pressure
- Ideally we want islands or nanoparticles to maximize metal-graphene-gas boundaries





Functionalization with metal and metal oxide nanostructures for selectivity tuning

Scanning Kelvin probe microscopy: Maps surface morphology and surface potential

> $\Delta \Phi$ between 1LG and 2LG allows nanoscale mapping of graphene thickness (and doping)



□ Morphology shows deposition of continuous porous metal – ideally: islands...

- □ 1LG/2LG potential contrast: surface retains the electronic properties of graphene
- □ Pt wets the surface better than Au screens the graphene for 'thick' depositions



As-grown graphene

Au decorated graphene

Response Time (min), 50 ppb NO₂

Pt, 5 nm

2.3

10.9

41.7

As-grown

316

834

2136

Au, 5 nm

1.5

9

74



General Effects of metallization:

- Improved speed of response
- Improved detection limit
- More stable base line
- Suppressed response to H₂/CO while maintaining NO₂ response (Au < 5 nm)

Response %

30%

60%

90%

J. Eriksson, D. Puglisi, Y. H. Kang, R. Yakimova, A. Lloyd Spetz, Physica B 439, 105–108 (2014)

As-grown

6

23

99



Pt, 2 nm

14,8

49

175,5

Recovery Time (min)

Au, 5 nm

14

47

135

Increased sensitivity



□ Detection limit < 1 ppb



Porous metal grains or nanoparticles increase the probability of interaction between the graphene surface and adsorbates



Designed Nanoparticles by Pulsed Plasma

It is expected that decoration with different metals or metal-oxide nanostructures will allow careful targeting of selectivity to specific molecules

Plasma-based nanoparticle (NP) synthesis process

Highly reproducible thin film deposition technique



Gas aggregation zone

Gas flow





□ The effect depends on the size of the deposited NPs (< 5 nm, sensitive to benzene, > 50 nm, sensitive to formaldehyde)





Graphene

CONCLUSIONS

- Sensing with epitaxial graphene promising, ppb level NO₂ detection
- Obstacles (selectivity and speed) are being overcome
- Thin (0.5 5 nm), porous decoration can result in improved selectivity, sensitivity, stability, and response/recovery times
 - The effect depends on the choice, thickness, and nanostructure of the decoration
- > Air quality control: ppb level detection limit required, a likely application
- Emerging interest in detection of VOCs in living environments ppb level detection crucial. Graphene is an excellent candidate
- It is expected that decoration with different metals or metal-oxide nanostructures will allow careful targeting of selectivity to specific molecules

