European Network on New Sensing Technologies for Air Pollution Control and Environmental Sustainability - *EuNetAir* COST Action TD1105 2<sup>nd</sup> International Workshop *EuNetAir* on *New Sensing Technologies for Indoor and Outdoor Air Quality Control* ENEA - Brindisi Research Center, Brindisi, Italy, 25 - 26 March 2014

### DETECTION OF LOW CONCENTRATIONS OF VOLATILE ORGANIC COMPOUNDS WITH SiC-FETs



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### Division of Applied Sensor Science and Research collaborations



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660		
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EURC	DPEAN COOPERATION IN SCI	ENCE AND TECHNOLOGY 2

### **Outline**

#### Motivation

- Why SiC-FET for indoor AQC
- Issues: evaluation/optimization of sensors' performance and characteristics
  - Sensitivity
  - Stability
  - Detection limit
  - Response/recovery time
  - Temperature dependence
  - Effect of relative humidity
- Main achievements and open problems



# **Motivation**





Air Quality

Adequate **control** of emissions **for** more efficient **reduction** of hazardous air pollutants

Development of **highly-sensitive low cost** gas sensors able to detect **ppb** concentrations of specific VOCs for **indoor** AQC.

# Why SiC-FET sensors?

- Wide band gap
- Chemical inertness of SiC



#### HIGH-TEMPERATURE OPERATION

#### **ADVANTAGES**

Possibility to operate the sensor over a wide temperature range with high, stable, reproducible performance

Flexibility when using temperature cycling mode and also possibility to use high temperature for regeneration of the sensor surface

Manufacturing based on proven semiconductor processes allowing costefficient mass production

Large experience on FET technology and high performance commercial transistor devices available





## SiC-FET gas sensors fabrication

The SiC-FETs are supplied by SenSiC AB, SME company, and EuNetAir partner. Different sensing layers and operation temperatures enables detection of H<sub>2</sub>, NH<sub>3</sub>, CO, NO<sub>x</sub>, HC like VOCs, H<sub>2</sub>S, SO<sub>2</sub>

- 30 nm Ir total thickness
- 300 µm gate width
- 10 µm gate length
- 5 μm separation between gate and source/drain





4" wafer ~ 1800 chips



## **FE Sensor platform**







Gate composed by a porous catalytic metal (Ir, Pt) as sensing layer

#### Sensitivity by

- Number of three phase
  boundaries gas-metal-oxide
- Adsorption sites on the insulator

#### Selectivity by

- Choice of temperature
- Different catalytic materials
- Structure of the metal

Gas adsorption/reaction at the gate contact => I-V shift

# Experimental



Gas mixing system at Saarland University, Laboratory for Measurement Technology, Germany

permeation

oven 1

permeation

oven 2

permeation ovens

MFC

MFC

gas dilution

MFC

20 ml/min

MFC 500 ml/min

MFC

humidification

synthetic air

synthetic air

formaldehyde

synthetic air

synthetic air





FORMALDEHYDE @ 330°C DRY AIR

Conc. (ppb)	Response (mV)	Response time (min)	Recovery time (min)
100 ppb	29	9.4	22.8
1 ppb	13	3.8	2.4
500 ppt	10	4.5	13.2
200 ppt	7	2.4	1.2





COST Action TD1105 EuNetAir Newsletter, Iss. 3/Dec. 2013.

#### NAPHTHALENE @ 330°C

Conc. (ppb)	Response (mV)		Response time (min)		Recovery time (min)	
	DRY AIR	10% R.H.	DRY AIR	10% R.H.	DRY AIR	10% R.H.
90 ppb	274	46	2.4	0.8	6.0	6.6
<b>10 ppb</b>	140	26	10.2	6.6	4.8	3.6





#### BENZENE @ 330°C

Conc. (ppb)	Response (mV)		Response time (min)		Recovery time (min)	
	DRY AIR	10% R.H.	DRY AIR	10% R.H.	DRY AIR	10% R.H.
7	21	9	5.6	2.4	10.8	4.8
5	18	9	3.9	n.a.	11.8	16.0
3	12	9	8.3	7.9	n.a.	9.5



## **Temperature dependence and** effect of relative humidity





D. Puglisi, J. Eriksson, C. Bur, A. Schuetze, A. Lloyd Spetz, M. Andersson, Mat.Sci. Forum 778-780 (2014) 1067-1070. RATION IN SCIENCE AND TECHNOLOGY

### CONCLUSIONS



# Financial support / Partnership





Basis results obtained during a two-week **STSM** at Saarland University (20 May - 4 June 2013)

Grant agreement no. 604311-2, start January 2014





Travel grant from KAW Foundation, IMCS in Buenos Aires (16-19 March 2014)

#### **Contributions to EuNetAir events available at: http://sensindoor.eu/publications/**

